

IDENTIFICATION OF ISOMERIC DIHYDRO-AZINES BY BENZENE
INDUCED SHIFTS IN NMR SPECTRA. VI¹

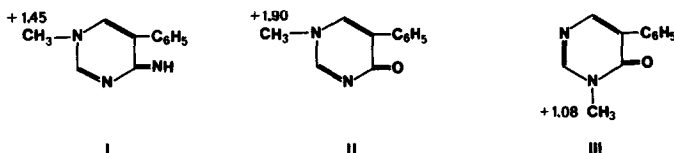
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It is well known that alkylation of 4-amino- or 4-hydroxy -pyrimidines may give two isomeric 1,4- or 1,6-dihydro-N-alkyl derivatives, which can be identified by chemical or spectroscopic methods². Of special interest is the determination with nmr spectroscopy based either on the chemical shifts³ or on the coupling constants of pyrimidine protons^{4,5}. However, we have found that the solvent shifts induced by benzene on nmr spectra can successfully be used for structure determination of isomeric dihydro-forms of N-alkyl azines.

The first compound studied was the methylation product⁶ of 4-amino-5-phenyl-pyrimidine, which was identified, by chemical means, as 1,4-dihydro-1-methyl-4-imino-5-phenyl-pyrimidine (I). The nmr spectrum of I in benzene showed for the N-CH₃ protons an unusually large diamagnetic shift $\Delta_{\text{C}_6\text{H}_6}^{\text{CDCl}_3}$ equal to 1.45 p.p.m.



Solvent shifts measurements made on the pair of 1,4-dihydro- and 1,6-dihydro-pyrimidone (II,III) showed that there is a dramatic difference on the shifts of methyl protons in II and III, $\Delta_{\text{N-CH}_3}$ being +1.90 and 1.08 p.p.m. respectively.

Examination of some other isomeric pairs analogous to II-III showed that the diamagnetic shift induced by benzene on N-alkyl protons is much higher in 1,4-dihydro-derivatives than it is in 1,6-, the corresponding shifts $\Delta_{\text{N-CH}_3}$ being 1.17-1.90 and 0.45-1.08 p.p.m. This method can easily and conclusively be used for structure

determination of various isomeric dihydro-azines.

The solvent shifts induced by benzene relative to CDCl_3 ($\Delta = \delta_{\text{CDCl}_3} - \delta_{\text{C}_6\text{H}_6}$ p.p.m.) of the compounds studied are reported in Table 1.

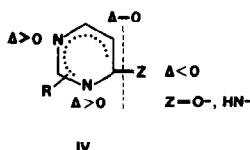
TABLE 1

Induced shifts by benzene relative to CDCl_3 ($\Delta = \delta_{\text{CDCl}_3} - \delta_{\text{C}_6\text{H}_6}$ p.p.m.) on various dihydro-azines.

Compound	$\Delta_{\text{N-alkyl protons}}$		Compound	$\Delta_{\text{N-alkyl protons}}$	
	a*	b*		a*	b*
I	1.45		1,4-Dihydro-1-methyl-pyridone-4	1.32	
II	1.90				
III		1.08	1,2-Dihydro-1-methyl-pyridone-2		0.58
1,4-Dihydro-1-methyl-4-imino-pyridine	1.17		1,2-Dihydro-1-methyl-2-imino-pyrimidine		0.62
1,2-Dihydro-1-methyl-2-imino-pyridine		0.45	1,4-Dihydro-1-ethyl-4-imino-pyridine	0.75 CH_2	
				0.58 CH_3	
			1,2-Dihydro-1-ethyl-pyridone-2		0.33 CH_2
					0.27 CH_3

*a, b Dihydro-azines analogous to II and III.

The observed difference in solvent shifts in various dihydro-forms (II-III) could be explained on the basis of the carbonyl plane rule^{7,8}. In this case the 1,4-dihydro-form should be expected to exhibit a larger shift than the 1,6-isomer (IV). Furthermore, in the latter case a steric effect might be operated in the "collision-complex" causing a smaller shift in alkyl protons.



It should be also noticed that the shifts induced by benzene are larger in oxo-derivatives than they are in imino-.

The nmr spectra were obtained with a Varian A-60A spectrometer at $\sim 40^\circ$, with TMS as an internal standard. The measurements were made in 2% solutions (w/v) in CDCl_3 and C_6H_6 . The solvent shifts Δ are given in p.p.m. The compounds studied were prepared by known procedures; their physical constants were in agreement with those of the literature.

R E F E R E N C E S

1. Part V, to appear in Org. Magn. Res.
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